1&3 Educational, Inc. Gateway to K

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MSDS No. Effective Date

SS 550 June 10, 1998 Threshold Limited Value

Effects of Overexposure

Emergency and

SECTION VI

Incompatibility

(Materials to Avoid)

Stability

First Ald Procedures

Unstable

Stable

Χ.

SECTIONI NAME 24 HOUR EMERGENCY ASSISTANCE Product SODIUM HYDROXIDE CHEMTREC Chemical Health 3 800-424-9300 Caustic Soda; Lye Synonyms Day 718-228-8177 Fire 0 Formula NaOH Reactivity 2 NEPA Unit Size up to 2.5 Kg. HAZARD RATING HMIS SLIGHT MODERATE HIGH EXTREME C.A.S. No. 1310-73-2 3 SECTION II INGREDIENTS OF MIXTURES Principal Component(s) **TLV Units** Sodium Hydroxide 90-100% See Section V. DANGERI CORROSIVEI POISON CAUSES SEVERE SKIN AND EYE BURNS. MAY BE FATAL IF SWALLOWED. DO NOT INHALE AS DUST OR MIST. SECTION III PHYSICAL DATA Melting Point (°F) 318°C (804°F) Specific Gravity (H.O = 1) 2.130 at 25°C Bolling Point (°F) Percent Volatile 1390°C (2534°F) Negligible as solid. by Volume (%) Evaporation Bate Vapor Pressure (mm Hg) < 20°C Non-volatile (NA), **~1)** Vapor Density (Air=1) N/A Solubility in Water 109 grams dissolves in 100 mL. water at 20°C. Appearance & Odor White pellets, flakes or beads; no odor. SECTIONIV FIRE AND EXPLOSION HAZARD DATA Flash Point Flammable Limits in Air Upper (Method Used) % by Volume Non-combustible (NA) Extinguisher Use water spray on fire involving this material. Media SPECIAL FIREFIGHTING In lire conditions, wear a NIOSH/MSHA-approved self-**PROCEDURES** contained breathing apparatus and full protective clothing. Must include complete eye protection. Flood with water, using care not to splatter or splash this material. Contact with water produces intense heat and highly imitating and corrosive mist. (1993 EMERGENCY RESPONSE GUIDEBOOK, DOT P 5800.6, GUIDE PAGE NO. 60)

UNUSUAL FIRE AND **EXPLOSION HAZARDS**

Not combustible but solid form in contact with moisture or water may generate sufficient heat to ignite combustible materials. Contact with most metals can generate hydrogen gas. Hot or molten material will react violently with water liberating heat and causing splashing. A severe eye hazard; solid or concentrated solution destroys tissue on contact.

D.O.T.	SODIUM HYDROXIDE, SOLID, 8, UN 1823, PG II	
Approved I	y U.S. Department of Labor "essentially similar" to form OSHA-2	0

Hazardous May form sodium monoxide and/or sodium peroxide at very high **Decomposition Products** temperatures. Hazardous Polymerization Conditions to Avoid May Occur Will Not Occur Not applicable. SECTION VII SPILL OR LEAK PROCEDURES Steps to be taken in case Carefully and wearing protective clothing, sweep up and place in material is released or spilled a sullable container. Flush spill area with water, rinse with dilute acid, preferably acetic, and finally with water. Discharge, treatment, or disposal may be subject to Federal. State or Local laws. Waste Disposal Method These disposal guidelines are intended for the disposal of catalog-size quantities only. Avoid breathing dust or mist. Wear full protective clothing including goggles or face shield. Slowly dissolve spill in water. While making solution add slowly to surface of stirred liquid to avoid violent splattering. Neutralize with sodium bisulfate and flush to sewer with copious amounts of water. SECTION VIII SPECIAL PROTECTION INFORMATION Respiration Protection None required in normal laboratory handling. If dusty conditions prevail, use a high (Specify Type) elliciency particulate respirator, Local Exhaust Recommended. Special No. Ventilation Mechanical (General) Other Recommended Chemical safety goggles, or **Protective Gloves** Eve Protection Rubber. face shield where appropriate Other Protective Goggles, lab coat, apron, ventilation hood, proper gloves, eye wash station. Equipment **SECTION IX** SPECIAL PRECAUTIONS Precautions to be Taken Keep container lightly closed. Store in a cool, dry place; protect against moisture and water. Separate from acids, metals, explosives, organic in Handling & Storing peroxides and easily ignitible materials. Avoid contact with skin, eyes and Keep container tightly closed when not in use. clothing. Wash thoroughly after handling. Reed label on container before using. Do not west contact lenses when working with chemicals, Other Precautions Product is deliquescent and absorbs water and carbon dioxide from air. Sodium hydroxide and trichloroathylene are especially hazardous since they react to form spontaneously flammable dichloroacetylene. Wash contaminated clothing promptly. For laboratory use only. Not for drug, food or household use. Keep out of reach of children. Revision No. 6 Date 08/10/88 Approved Michael Raszela The information contained herein is furnished without warranty of any kind. Employers should use this information only as a supplement to other information gethered by them and must make independent determinations of sulfability and completeness of information from all givenes to assure proper use of these netadale and the calety and hackle of amelinians . Blazanhiso Blatailala Indistribat Standards . Pilitad on & manar

TLV/TWA: 2 mg/m3 (Air) (Celling value, as 100%)

INGESTION: May result in severe intestinal initiation with burns to mouth, throat and stomach with nauses and vomiting. SKIN AND EYES:

Contact with skin or eyes may cause severe inflation or burns. INHALA.

TION: Severe Initiation to respiratory system with pulmonary edema, jung

INGESTION: If swallowed, do NOT Induce vomiting. If conscious, give

large amounts of water to drink. Follow with white of eggs, beaten with

water. Call physician immediately. Never give anything by mouth to an unconscious person. EYES: Immediately flush with large amounts of

water for 15 minutes, lifting lower and upper eyelids occasionally. Get

prompt medical attention. SKIN: Flood with water, then wash with

Can react violently with acids and with many organic compounds. Reacts with most

common metals (zinc, aluminum, tin, lead, etc.) liberating flammable hydrogen gas.

vinegar. INHALATION: Remove to fresh air. If not breathing, give

artificial respiration. If breathing is difficult, give oxygen. Get medical

Moisture, acids and acid fumes.

(ACGIH 1992-93).

inflammation.

attention.

REACTIVITY DATA

Conditions to Avoid